

TOTAL SYNTHESIS OF DIOSPYROL AN ANTHELMINTIC DRUG FROM *DIOSPYROS MOLLIS* GRIFF

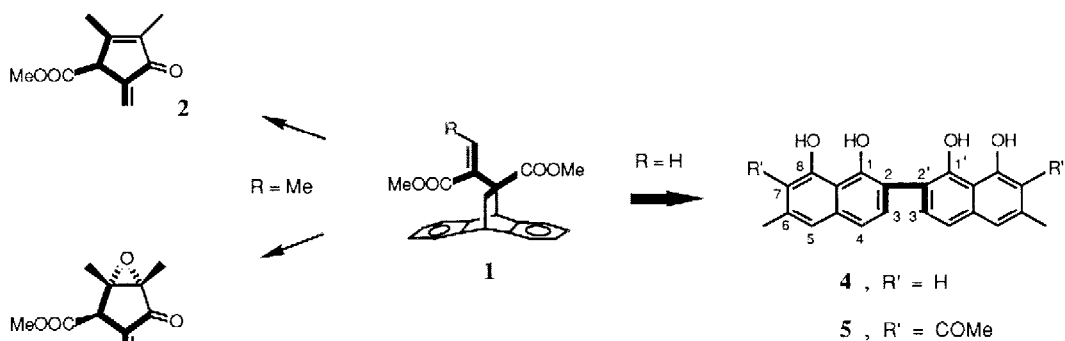
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Abstract : Diospyrol **4** is constructed from a masked butadiene synthon, anthracene adduct **1** [R = H]. The key construction steps comprise tandem Michael addition-Dieckmann condensations, first between anion **7** and **1**, and subsequently between anion **7** and **15**.

In the previous communication ¹ we described a straightforward synthesis of deepoxy-4,5-didehydromethylenomycin A and methylenomycin A methyl esters, **2** and **3** respectively, from anthracene adduct **1** [R = Me]. To further elaborate the usefulness of anthracene adducts we report here a short and convenient total synthesis of **Diospyrol 4**, starting from **1** [R = H].

Scheme I

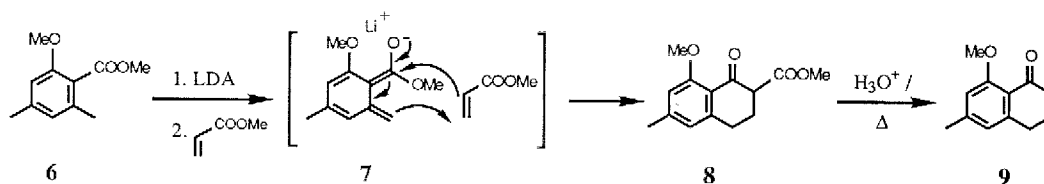


Diospyrol, a dimeric naphthol derivative with alleged anthelmintic property, is obtained from the berry of *Diospyros mollis*,² an indigenous plant widely distributed throughout Southeast Asia. The *Diospyros* berry has been extensively used in Thailand for treating hook-worms since ancient times, until suddenly, over a decade ago, cases of blindness were reported in a mass chemotherapy programme conducted by the Ministry of Health, which threw confusion and uncertainty

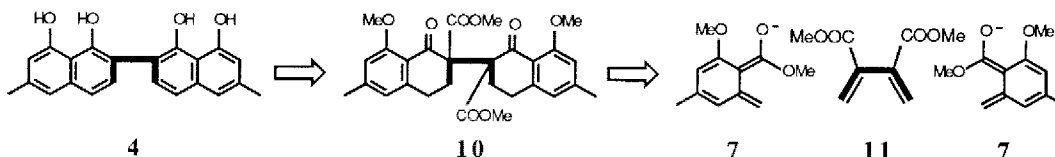
over the continued use of the berry. Coincidentally a toxic compound, **stypanrol** or "blindgrass toxin" **5** [isolated from an Australian plant *Stypandra imbricata* or "blindgrass"]³ which has a similar structure to diospyrol **4**, is reported to cause blindness in animals when ingested. Naturally it became of interest to synthesize diospyrol **4**, stypanrol **5**, and their derivatives for a study of structure-reactivity relationships.

A former synthesis of diospyrol, reported by Govindachari,⁴ utilized classical manipulations which gave a poor overall yield. An improved method, recently developed by the Australian group,⁵ delivers stypanrol **5** via Fries rearrangement of 1,1'-dimethoxy-8,8'-diacetyldiospyrol followed by demethylation. Our synthesis evolved from our prior observation that 8-methoxy-1-tetralones, e.g. **9**, can easily be obtained in a one-pot process from the reaction between toluate anion **7**⁶ and acrylic ester as shown in **Scheme II**.⁷ Using a retro-synthetic approach, we dissected symmetrical diospyrol **4** into a central four carbon synthon [drawn in heavy line in **Scheme III**] and identical left and right ring synthons. Since anthracene adduct **1** yields 2,3-dicarbomethoxybutadiene **11** upon flash vacuum pyrolysis,⁸ it nicely provides the requisite four central carbon unit upon which to construct the left and right naphthol rings by the annulation reaction of the type shown in **Scheme II**.

Scheme II



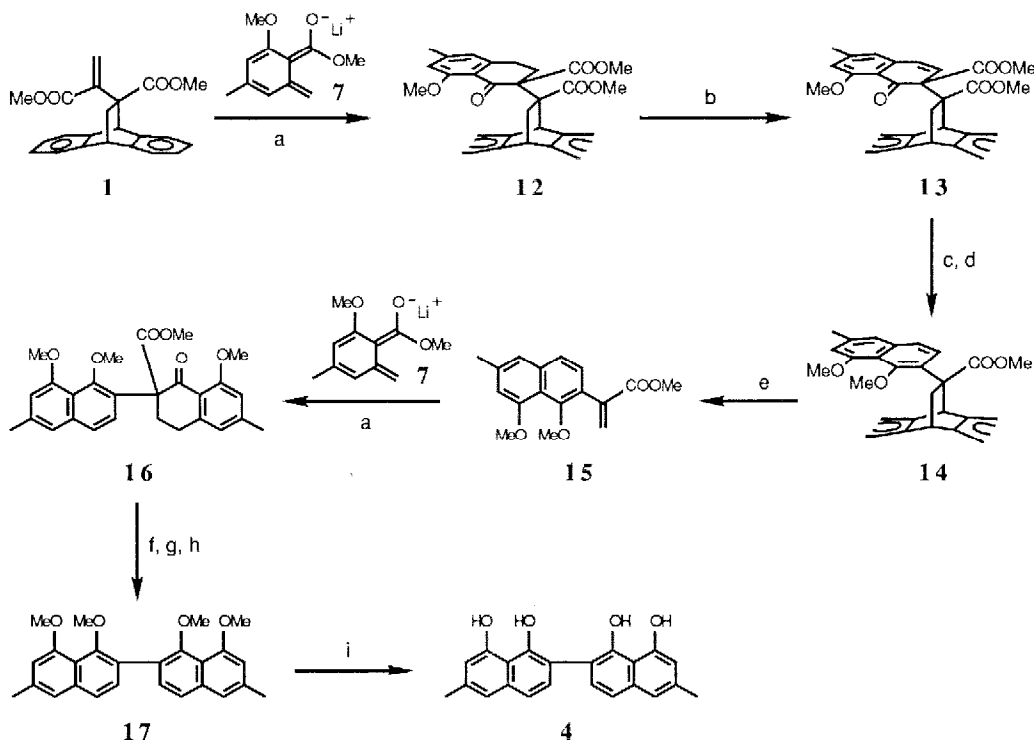
Scheme III



Addition of diene **11** [1 eq.] to anion **7** [2 eq.] in THF solution at -78° with prolonged stirring at various temperatures consistently gives complex mixtures. However, treating the diene precursor **1** [R = H] with an equimolar amount of **7** at -78° and leaving the reaction mixture to stir overnight at room temperature cleanly yields **12** as a mixture of two separable isomers [**12a** and **12b**, ratio 1:2; silica gel, 25% ethyl acetate in hexane as eluant; 80%]. Both isomers fail to undergo the hydrolysis-decarboxylation reaction under the acidic conditions earlier employed,⁷ but can be dehydrogenated with DDQ in boiling dioxane to give **13a** (60% from **12a**) and **13b** (50% from **12b**) respectively. Compounds **13a** and **13b** can then be hydrolysed, decarboxylated and converted into the 1,8-dimethoxynaphthalene derivative **14** [colourless solid mp. 187-189° from a dichloromethane-hexane mixture; 48 and 52% from **13a** and **13b**

respectively]. Flash vacuum pyrolysis of **14** gives analytically pure vinylnaphthalene **15** in 92% yield. Its nmr spectrum clearly indicates the presence of an aromatic methyl at δ 2.44, three methoxy groups [δ 3.72, 3.78, and 3.94], two vinylic protons at δ 5.77 and 6.32 [$J = 2$ Hz], and four aromatic protons as two doublets [δ 6.68 and 7.16; $J = 1.5$ Hz] and an AB quartet at δ 7.22 and 7.47 [$J = 9$ Hz].

Scheme IV



a) THF / -78° \rightarrow r.t. overnight ; b) DDQ / dioxane / reflux ; c) 30% aq. NaOH / EtOH / THF / RT ; d) Me_2SO_4 / K_2CO_3 / acetone / reflux ; e) 500° / 0.01 mm ; f) 30% NaOH / dioxane-MeOH / Δ then H_3O^+ ; g) $(\text{MeO})_3\text{CH}$ / TsOH / MeOH / Δ ; h) DDQ / benzene / r.t. ; i) BCl_3 / CH_2Cl_2 / -60° \rightarrow RT

With **15** in hand, the simple task remaining is to repeat the annulation sequence once more. Thus treatment of **15** with **7** as previously described gives rise to **16** [74%] which, after hydrolysis and decarboxylation, is converted into the corresponding enol ether **9** and finally dehydrogenated with DDQ to yield diospyrol tetramethylether **17** [67%] ¹⁰ which possesses physical properties identical in all respects with an authentic sample. Boron trichloride demethylation **4** of **17** completes the synthesis of the target diospyrol **4** in 85% yield. However, because diospyrol is rather unstable and can be easily air-oxidised, it is preferable to store the compound in the form of its tetramethylether **17**, especially for

transportation purposes. Diospyrol can then be freshly generated just before use. Various derivatives of **4** are currently being synthesized for biological testing.

The use of adduct **1** in the above synthesis represents an unconventional mode of construction of the aromatic naphthol nucleus, i.e. via a tandem Michael addition-Dieckmann condensation process instead of the customary electrophilic substitution. The synthesis of diospyrol outlined above thus serves to emphasize the versatility of anthracene adducts in organic synthesis.

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A preliminary account of this work was presented at the 22nd Sheffield Stereochemistry Symposium , 14 December 1988.

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10. Application of the f, g, h reaction sequence [see conversion of **16** \rightarrow **17**] to **12** gave a much lower yield of **14** than that obtained by the described b, c, d route.

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